M1.(a) Heating <u>speeds up</u> (hydrolysis / breaking of peptide bonds)

OR forms non-sweet (amino acids)

(b) (2-)aminobutan<u>e</u>dioic acid OR 2 not necessary but penalise other numbers at start

(2-)aminobutane(-1,4-)dioic acid

1,4 not necessary but penalise other numbers and 1,4 must be in correct place (QoL)

H₂N-C-COO H₂N-C-COO CH₂ (c) COO⁻

allow -CO,allow NH₂-

 $CH_2 - C - H$ $COO^$ *allow* $-CO_2^$ *allow* $+NH_3$ *don't penalize position of* + *on* NH_3

1

1

1

1

1

- (e) (i) **M1** Compounds/molecules with same <u>structural formula</u> *Not just structure*
 - M2 But with <u>bonds/atoms/groups</u> arranged differently <u>in space</u> or <u>in 3D</u> *Allow –with different <u>spatial</u> arrangement of <u>atom/bond/group</u> 1*

Independent marks





1

1

4

1



(b) <u>2-bromobutanenitrile</u> Allow 2-bromobutane-1-nitrile

(c) M1 ammonia or NH₃

M2 excess (ammonia) excess tied to NH₃ and may score in M1 unless contradicted Ignore concentrated or sealed container, Acid loses conditions mark

M3 nucleophilic substitution Allow close spelling

(ii) **M1** electrostatic <u>forces between ions</u> in **X QOL** Allow ionic bonding.

Marks independent

 M2 (stronger than) <u>hydrogen bonding</u> between CH₃CH₂CH(OH)COOH
 CE mention of molecules of X or inter molecular forces between X loses both marks

1

1

1

1

1

COOCH2CH3OR COOCH2CH3 (e) (i) Isomer of C₄H₉NO₂



(iii) $H_2N-CH_2CH_2CH_2-COOH$ or $H_2N-(CH_{2)3-COOH}$ Isomer of $C_4H_9NO_2$ allow NH_2-

OR

 $(f) \qquad \begin{array}{c} \mathsf{N}(\mathsf{CH}_3)_2 \\ \mathsf{CH}_3\mathsf{CH}_2 & -\mathsf{C} & -\mathsf{H} \\ \mathsf{(f)} & \mathsf{COOH} \\ & Answer has \ 6 \ carbons \ so \ \textbf{NOT} \ isomer \ of \ \textbf{X} \\ & Allow \ C_2H_s \\ & Must \ have \ bond \ from \ \mathsf{C} \ to \ \mathsf{N} \ not \ to \ methyl \ group \end{array}$

[16]

1

1

1

1

1

M3. (a) polyamide or nylon (2,4) (allow nylon without numbers but if numbers are present they must be correct) condensation

(b)
$$H_3 N - CH_2 - COO$$

(c) ionic bonding in aminoethanoic acid (can only score if includes that aminoethanoic is ionic)

stronger attractions than Hydrogen bonding in hydroxyethanoic acid (e.g. stronger Hydrogen bonding in aminoethanoic acid scores 0) (mention of electrostatic forces between molecules scores 0)

M4. (a) (i) $H_{3}^{+} H_{C}^{-} COO^{-}_{CH_{3}}$ *allow* $-CO_{2}^{-}_{allow 'NH_{3}^{-}}$

allow -CO2 allow 'NH3don't penalize position of + on NH3



1

1

1

1

1

1

[5]



(b)



allow $-CO_2H$ allow NH_2 allow C_3H_7 allow as zwitterions if error in peptide link e.g. O H

if twice, penalise both times not polymers

if wrong amino acid in both can score Max 1

(c) chromatography or electrophoresis ignore qualification to chromatography

[6]

1

1

1

not hydration

1

1

1

1

1

(iii)

$$H_{3}^{+}N \xrightarrow{CH(CH_{3})_{2}}H_{3}^{+}N \xrightarrow{C}COO^{-}$$

(iv)

$$H_{3}^{+}N - C - (CH_{2})_{4}NH_{3}$$

$$H_{3}^{+}N - C - (CH_{2})_{4}NH_{3}$$

$$H_{3}^{-}$$

(b) (i)



(ii)



allow $-CO_2H$ allow -CONH- or -COHN- allow NH_2-



M6. (a) 3-hydroxypropanoic acid allow 3-hydroxypropionic acid must be correct spelling

1

1

[6]

(b) (i) must show trailing bonds



or can start at any point in the sequence, e.g.





1







1





allow polyalkene conseq on their c(ii) ignore n

1







In (e), do not penalise a slip in the number of carbons in the -CH₂CH₂- chain, but all must be bonded correctly NB two carboxylate groups Allow COONa or COO- Na⁺ but not covalent bond to Na allow NH₂-

$$H_2N - C - CH_2 - CH_2 - COOCH_3$$

H₂N - C - CH₂ - CH₂ - COOCH₃
H

OR



1

(iii)



In 4(e), do not penalise a slip in the number of carbons in the $-CH_2CH_2$ - chain, but all must be bonded correctly allow anhydride formation on either or both COOH groups (see below) with or without amide group formation

