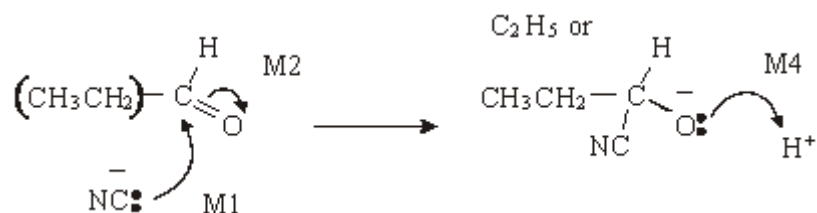


M1. (a) nucleophilic addition;



1

M3 structure;

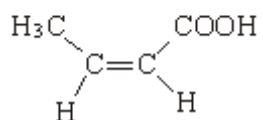
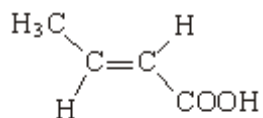
*(be lenient on position of charge on CN- )  
 (M2 not allowed independent of M1,  
 but allow M1 for correct attack on C+  
 if M2 show as independent first.)  
 (+on C of C=O loses M2 but ignore  $\delta+$  if correct)  
 (M4 for arrow and lone pair (only allow for correct M3 or  
 close))*

4

(b) (i) 2-hydroxybutanoic acid

1

(ii)

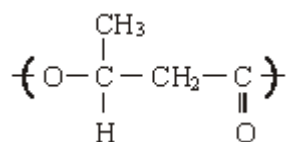


1

geometric(al) or cis-trans

1

(c) (i)



*(one unit only) (ignore brackets or n) (trailing bonds are*

needed)

1

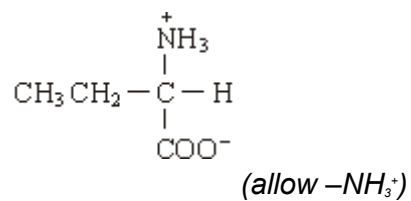
(ii) can be hydrolysed

OR

can be reacted with/attacked by acid/base/nucleophiles/H<sub>2</sub>O/OH<sup>-</sup>;

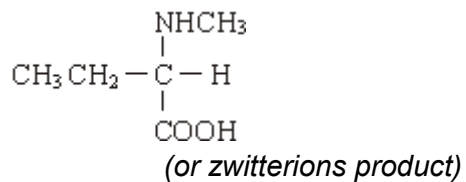
1

(d) (i)



1

(ii)



1

(iii) nucleophilic substitution;

1

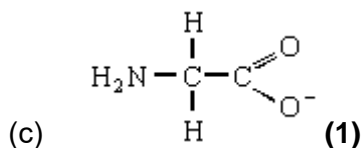
[14]

**M2.** (a) 2-amino(e) propanoic acid (1)

1

- (b) (i) molecules with same structure / structural formula **(1)**  
but with bonds **(atoms or groups)** arranged differently in  
space (3D) **(1)**
- (ii) Plane polarised light **(1)**  
Rotated (equally) in opposite directions **(1)**

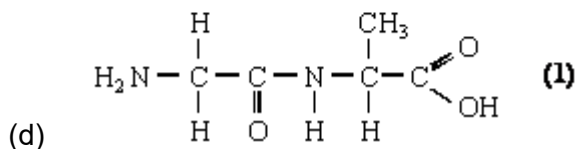
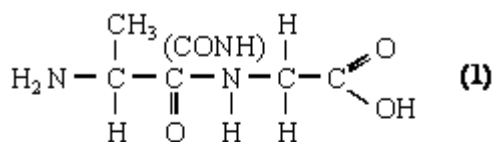
4



allow  $\text{H}_2\text{NCH}_2\text{COO}^-$

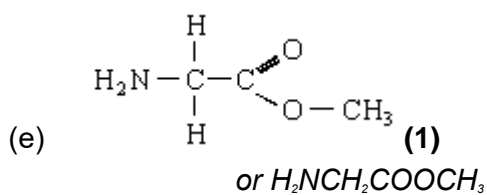
Penalise  $\text{NH}_2^-$  and  $\text{OH}^-$  once per paper  
but  $\text{CH}_3^-$  is allowed

1



Not anhydrides; not repeating units

2



1

[9]

M3. (a) 2-chloropropanoic acid (1)

1

(b)  $\delta$  1.72 Doublet  $\therefore$  next to CH (1)

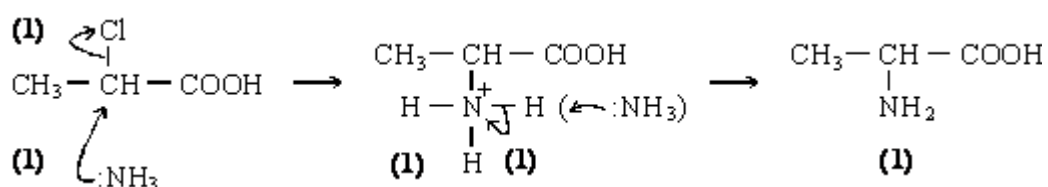
$\delta$  4.44 Quartet  $\therefore$  next to CH<sub>3</sub> (1)

2

(c) Two triplets (1)

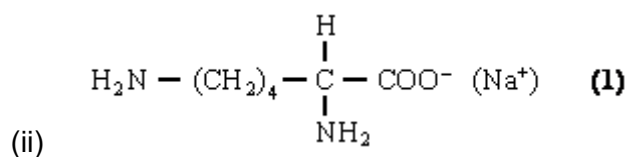
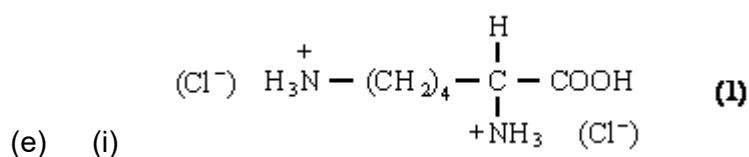
1

(d)

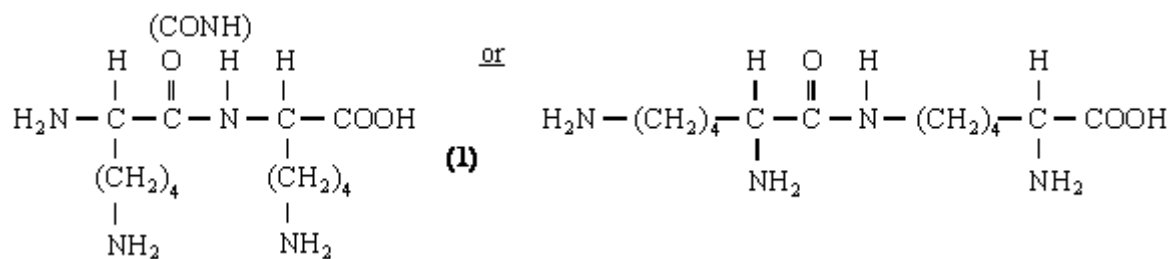


Allow S<sub>N</sub>1

5



(iii)

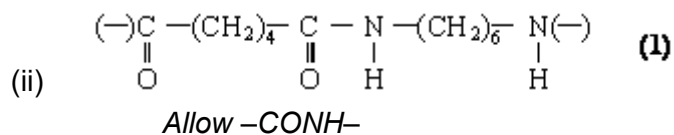


Or anhydride

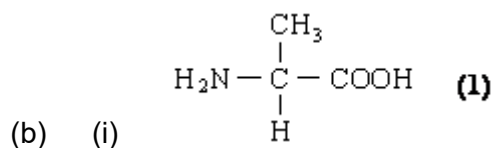
3

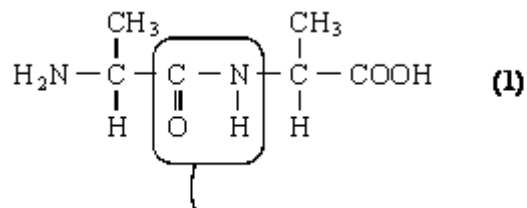
[12]

- M4.** (a) (i) hexane-1,6-diamine or 1,6-diaminohexane (**allow ammine**)  
or 1,6 hexan(e)diamine **(1)**



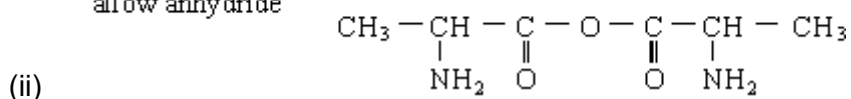
2





peptide link essential : the rest is consequential on b(i)  
(allow CONH)

allow anhydride



2

(c) (i) quaternary ammonium bromide salt (1)

(not ion, not compound)

Allow quaternery

(ii) Reagent:  $\text{CH}_3\text{Br}$  or bromomethane (1)

penalise  $\text{CH}_3\text{Cl}$  but allow excess for any halomethane

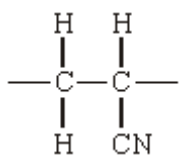
Condition: excess ( $\text{CH}_3\text{Br}$ ) (1)

(iii) nucleophilic substitution (1)

4

[8]

M5. (a) (i)



(Ignore n or brackets, but trailing bonds are essential)

1

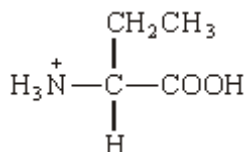
(ii) Addition or radical

1

(b) (i) 2-aminobutanoic (acid)

1

(ii)

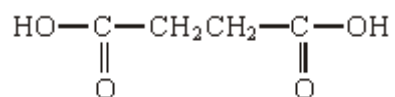


1

(c) (i)  $\text{C}_3\text{H}_4\text{O}_2$

1

(ii)



1

(1,4-)butan(e)dioic (acid)

*(allow succinic, but not dibutanoic nor butanedicarboxylic acid)*

1

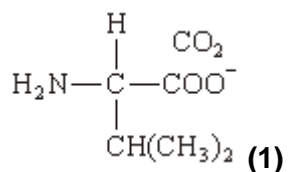
(iii) Can be hydrolysed / can react with acid or base or water /  
can react with nucleophiles

1

[8]

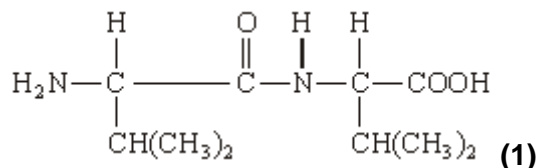
M6.

(a) (i)



*ignore  $\text{Na}^+$  unless covalently bonded*

(ii)



*must be dipeptide, not polymer nor anhydride*

*allow  $-\text{CONH}-$  or  $-\text{COHN}-$*

*allow zwitterion*

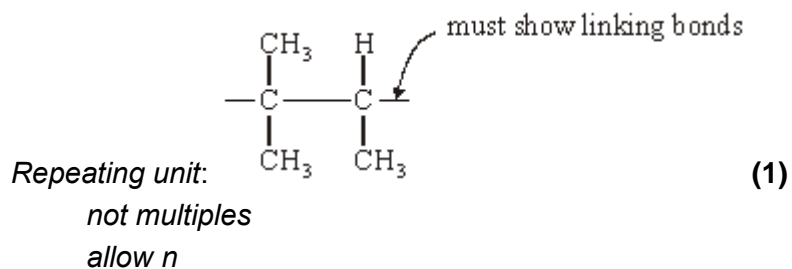
(iii) hydrogen bonding (1)

QL

Allow with dipole-dipole or v derWaals, but not dipole-dipole etc alone

3

- (b) (i) Type of polymerisation: addition(al) (1)



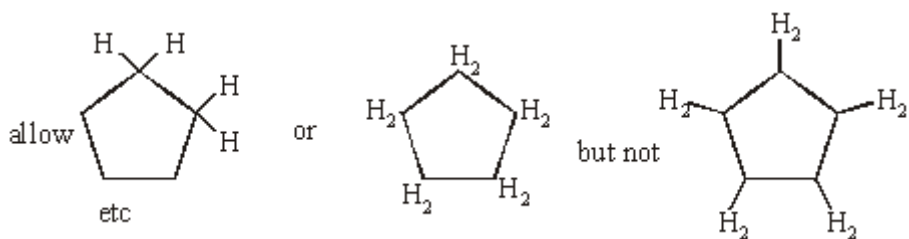
- (ii)  $\text{CH}_3\text{CH}=\text{CHCH}_2\text{CH}_3$  (1)  $\text{C}_6\text{H}_{10}$

- (iii)



(1)

or



4

[7]