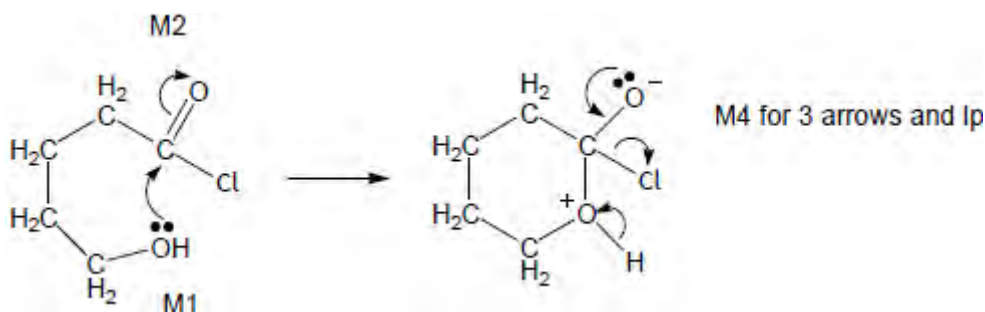


M1.(a) (i) (nucleophilic) addition-elimination

Not electrophilic addition-elimination

Ignore esterification

1



M3 for structure

- If wrong nucleophile used or O-H broken in first step, can only score M2.
- M2 not allowed independent of M1, but allow M1 for correct attack on C+
- + rather than $\delta+$ on C=O loses M2.
- If Cl lost with C=O breaking lose M2.
- M3 for correct structure with charges but lone pair on O is part of M4.
- Only allow M4 after correct / very close M3.
- Ignore HCl shown as a product.

4

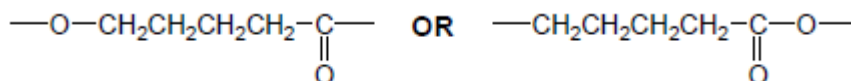
a 20-50 (ppm) or single value or range entirely within this range
If values not specified as a or b then assume first is a.

1

b 50-90 (ppm) or single value or range entirely within this range

1

(ii)



Must have trailing bonds, but ignore n.

1

Allow trough, peak, spike.

1

K Four (peaks)

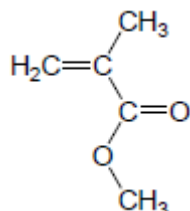
Ignore details of splitting.

If values not specified as J or K then assume first is J.

1

- (c) If all the structures are unlabelled, assume that the first drawn ester is L, the second ester is M; the first drawn acid is N, the second P. The cyclic compound should be obvious.

L
ester



OR $H_2C=C(CH_3)COOCH_3$

All $C_5H_8O_2$ L to P must have $C=C$.

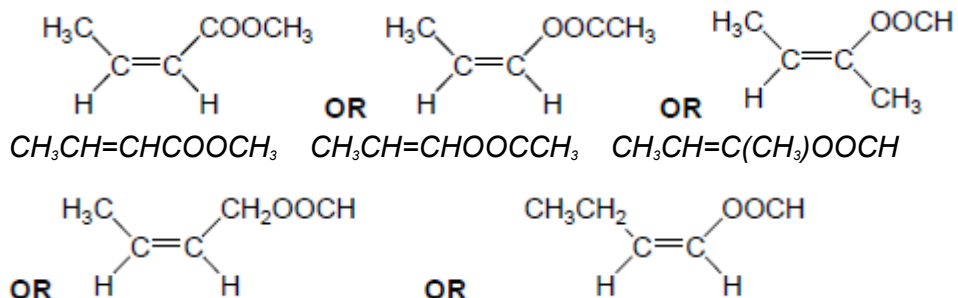
Allow CH_3- .

Allow $-CO_2CH_3$ etc.

Allow $CH_2C(CH_3)COOCH_3$.

1

M
ester



$CH_3CH=CHCH_2OOCH$

$CH_3CH_2CH=CHOOCH$

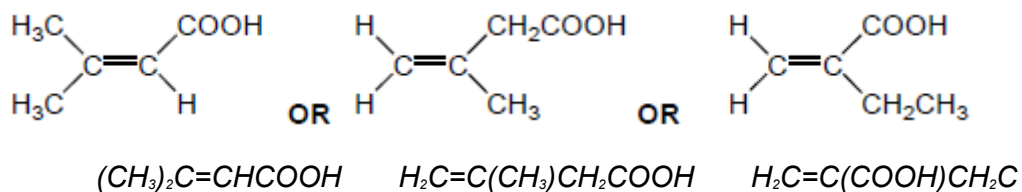
Allow either E-Z isomer.

Allow CH_3- or C_2H_5- but not CH_2CH_3- .

Allow $CH_3CHCHCOOCH_3$ etc.

1

N
acid



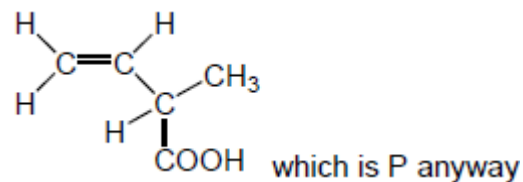
H_3

Allow CH_3 - or C_2H_5 - but not CH_2CH_3 -.

Allow $-\text{CO}_2\text{H}$.

Not cyclic isomers.

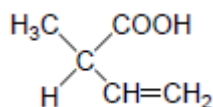
Not the optically active isomer.



Allow $(\text{CH}_3)_2\text{CCHCOOH}$ etc.

1

P
acid



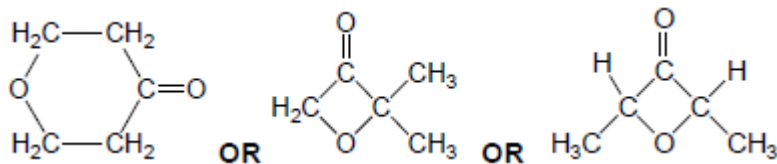
Allow $-\text{CO}_2\text{H}$.



Allow $\text{CH}_3\text{CH}(\text{CO}_2\text{H})\text{CHCH}_2$ or
 $\text{CH}_3\text{CH}(\text{CO}_2\text{H})\text{C}_2\text{H}_5$.

1


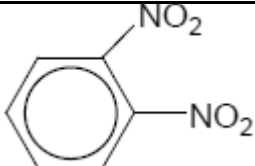
Q



Not cyclic esters.

1

[19]

M2.(a) F	G
	

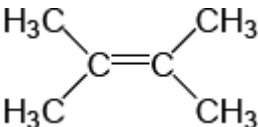
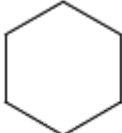
Penalize $-O_2N$ once

Penalise missing circle once

Don't penalise attempt at bonding in NO_2

1

1

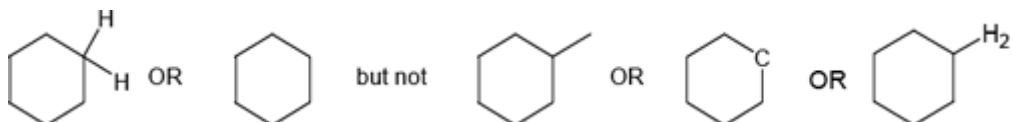
(b) H	J
	

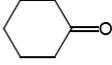
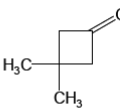
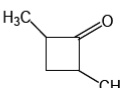
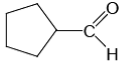
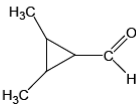
If **both H and J** correct but reversed, award one mark

1

1

A carbon in saturated ring structures should be shown as



(c) K	L
 OR  OR 	 OR 

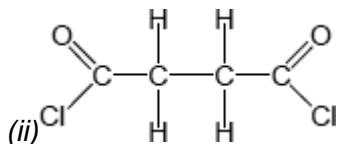
1

1

(b) (i) pentane-1,5-diol

Second 'e' and numbers needed

Allow 1,5-pentanediol but this is not IUPAC name



Must show ALL bonds

1

(iii) All three marks are independent

M1 (base or alkaline) Hydrolysis (allow close spelling)

1

Allow (nucleophilic) addition-elimination or saponification

M2 δ^+ C in polyester

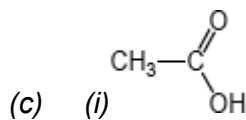
1

M3 reacts with OH⁻ or hydroxide ion

1

Not reacts with NaOH

1



Allow CH₃COOH or CH₃CO₂H

1

(ii) (nucleophilic) addition-elimination

Both addition and elimination needed and in that order

OR

(nucleophilic) addition followed by elimination

Do **not** allow electrophilic addition-elimination / esterification

Ignore acylation

1

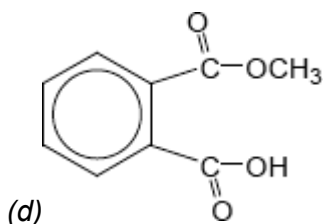
(iii) any **two** from: ethanoic anhydride is

- less corrosive
- less vulnerable to hydrolysis
- less dangerous to use,
- less violent/exothermic/vigorous reaction OR more controllable rxn
- does not produce toxic/corrosive/harmful fumes (of HCl) OR does not produce HCl
- less volatile

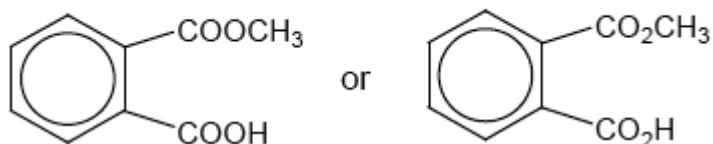
NOT COST

List principle beyond two answers

2



Allow



1

(e) (i) ester

Do **not** allow ether

Ignore functional group/linkage/bond

1

(ii) 12 or twelve (peaks)

1

(iii) 160 – 185

Allow a number or range within these limits

Penalize extra ranges given

Ignore units

1

(f)	(i)	sulfuric acid	sodium hydroxide	✓
		hydrochloric acid	ammonia	X or blank
		ethanoic acid	potassium hydroxide	✓
		nitric acid	methylamine	X or blank

4 correct scores 2

3 correct scores 1

2 or 1 correct scores 0

2

(ii) Pink to colourless

Allow 'red' OR 'purple' OR 'magenta' instead of 'pink'

Do **not** allow 'clear' instead of 'colourless'

1

[21]

M4.(a) (i) Single / one (intense) peak / signal **OR** all H or all C in same environment **OR**
12 equiv H or 4 equiv C

Do not allow non-toxic or inert (both given in Q)

Any 2 from three

Ignore peak at zero

OR

Upfield / to the right of (all) other peaks **OR** well away from others **OR**
doesn't interfere with other peaks

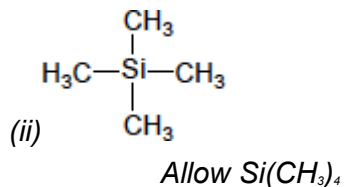
Ignore cheap

Ignore non-polar

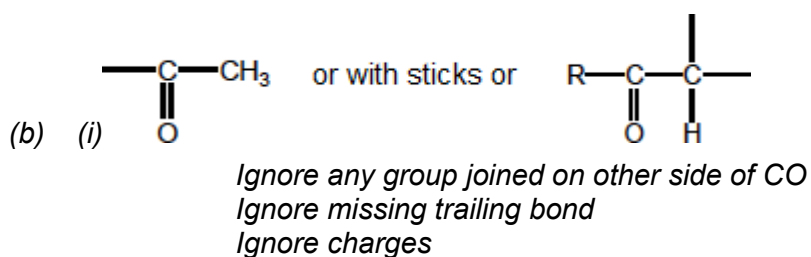
OR

Low bp **OR** volatile **OR** can easily be removed
Ignore mention of solubility

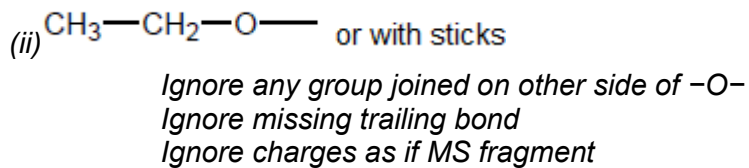
2



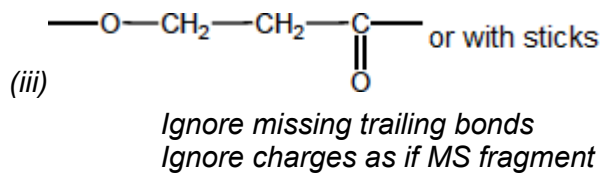
1



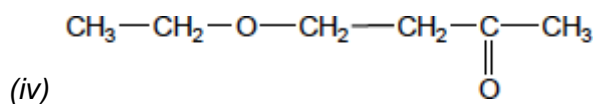
1



1

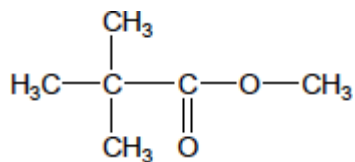


1



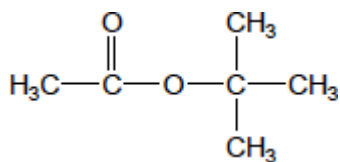
1

(c) (i) Check structure has 6 carbons



Allow $(\text{CH}_3)_3\text{CCOOCH}_3$ or $(\text{CH}_3)_3\text{CCO}_2\text{CH}_3$

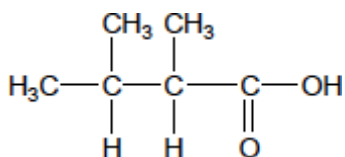
1



Allow $\text{CH}_3\text{COOC}(\text{CH}_3)_3$ or $\text{CH}_3\text{CO}_2\text{C}(\text{CH}_3)_3$

1

(ii) Check structure has 6 carbons

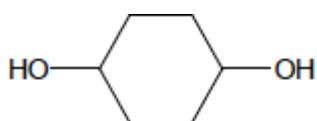


Allow $(\text{CH}_3)_2\text{CHCH}(\text{CH}_3)\text{COOH}$ or $(\text{CH}_3)_2\text{CHCH}(\text{CH}_3)\text{CO}_2\text{H}$

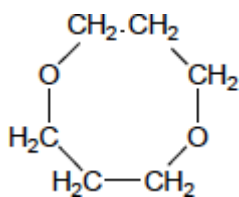
Penalise C_3H_7

1

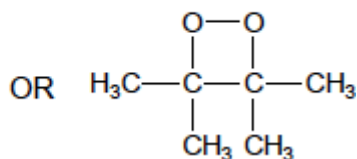
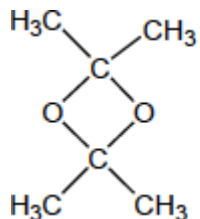
(iii) Check structure has 6 carbons



OR



Allow



1

[11]

