



# **A-Level Chemistry**

## **DNA**

### **Mark Scheme**

**Time available: 60 minutes**

**Marks available: 51 marks**

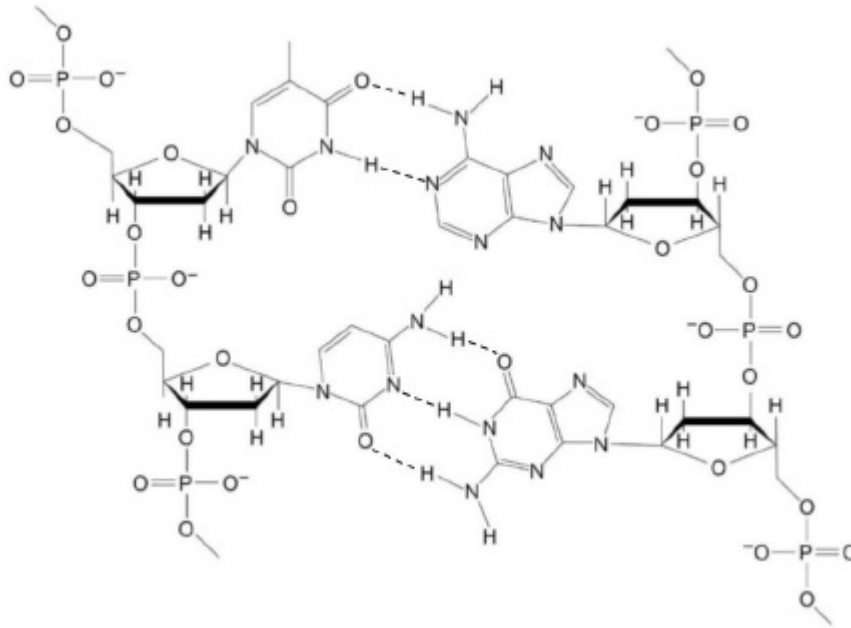
**[www.accesstuition.com](http://www.accesstuition.com)**



Mark schemes

1.

(a)



*M1* scored for the 2 H 'bonds' between A and T

1

*M2* scores for the 3 H 'bonds' between C and G

1

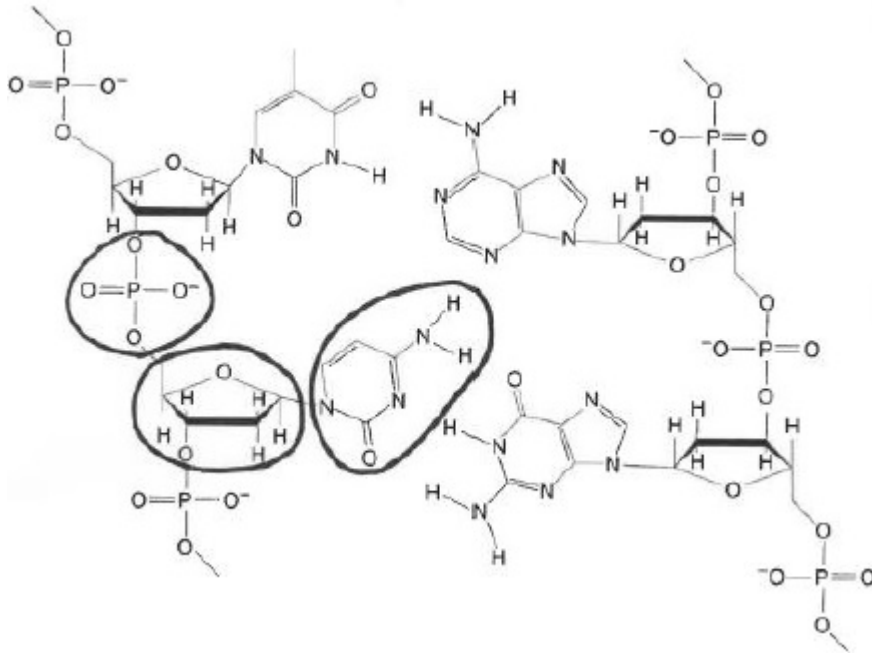
*Lose 1 for each extra 'bond'*

*H bonds must be linear*

*Penalise the use of full bonds instead of dashed lines once only*

*Ignore lone pairs and partial charges even if wrong*

(b)



**M1** scored for correct selection of cytosine and associated sugar

1

**M2** scored for selection of correct (upper) phosphate

1

**M1 & M2** can be scored with one 'ring'

Allow ring either side of the top O of either phosphate

If wrong base circled, can score **M2** for correct phosphate conseq to their base, i.e. top left, Thymine it's the upper phosphate top right, Adenine it's the lower phosphate bottom right, Guanine it's the lower phosphate

(c) (Complementary means the two strands must have base sequences) that match (all)  
A to T and C to G

Ignore reference to (hydrogen) bonding

1

[5]

2.

(a) 1 2 3 4 5

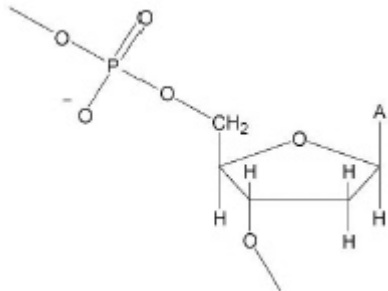
T G C A G

1

(b) 13

1

(c)



1 for completed 2-deoxyribose plus A

*Allow either OH or trailing bonds*

*Don't penalise 'sticks' in 2-deoxyribose.*

1

1 for correct phosphate joined to CH<sub>2</sub>

*If two phosphates shown CE=0*

*If CH<sub>2</sub> missing award 1 if no further errors*

*If phosphate attached to oxygen on C3 award 1 if no further errors*

1

[4]

3.

(a) X – base

1

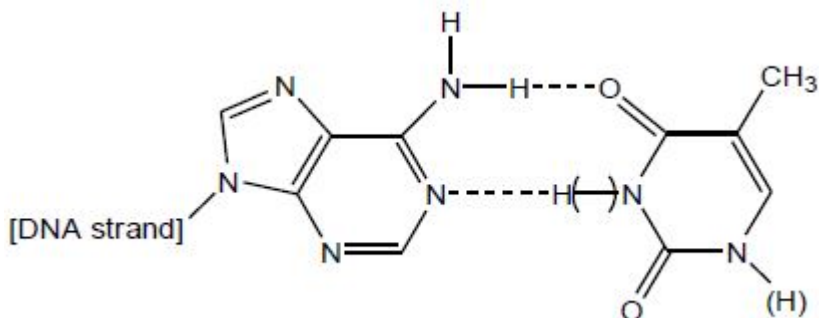
Y – phosphate (group)

1

*Ignore organic*

*Any mention of sugar in either loses that mark*

(b) If not Thymine CE=0



Correct structure scores 2, penalise by 1 each error in

- structure of thymine
- orientation of thymine
- hydrogen bonding

*Ignore lp on N and O*

*Don't penalise non-linear H bonds*

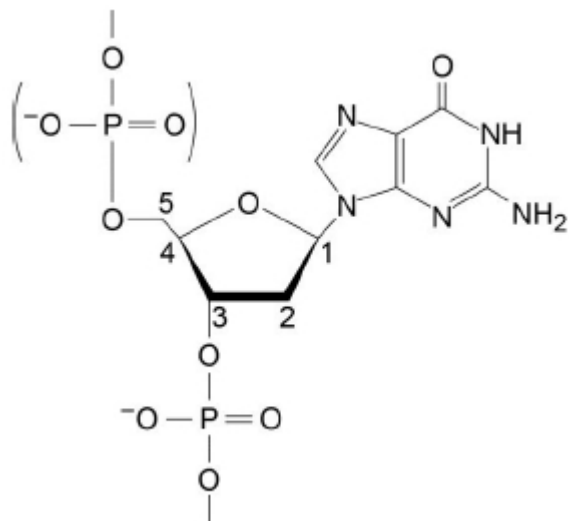
*on RHS of thymine – allow with or without H or – [DNA strand]*

2

[4]

4.

(a)



*CE=0 if **nucleotide** does not contain one base, one sugar and one phosphate.*

*Max 2 for any slips in structures.*

Correct phosphate-sugar link on C3.

*Allow phosphate attached to C5.*

1

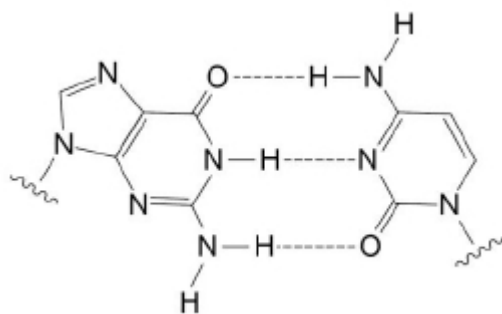
Correct sugar-guanine link on C1.

1

Remainder of molecule correct.

1

(b)



Correct diagram of cytosine (base pair with guanine).

*CE=0 if wrong base shown.*

1

Three hydrogen bonds drawn.

*Allow M2 if slip in M1.*

1

(c) There are only two H-bonds in the adenine-thymine base pair.

*Allow there is one fewer H-bond in the AT base pair.*

1

(d) The amino/-NH<sub>2</sub> groups in urea

1

are able to substitute for the H-bonds in the double helix.

*Allow H bonds will form between the urea and the DNA strands.*

1

**[8]**

**5.**

(a) DNA Replication

*NOT mitosis*

*NOT DNA synthesis*

*Ignore terms relating to cell division processes*

*Ignore 'damages DNA'*

*Ignore DNA transcription*

*Ignore 'cell replication'*

1

(b)  $[\text{Pt}(\text{NH}_3)_2\text{Cl}_2] + \text{H}_2\text{O} \rightarrow [\text{Pt}(\text{NH}_3)_2\text{Cl}(\text{H}_2\text{O})]^+ + \text{Cl}^-$

**M1** Correct formula **and** charge of B

1

**M2** Correct balancing **and** charges in equation

1

*Allow **M2** if the **only** error in complex B is the charge*

*(**M1** not awarded) with Cl<sup>-</sup> **only***

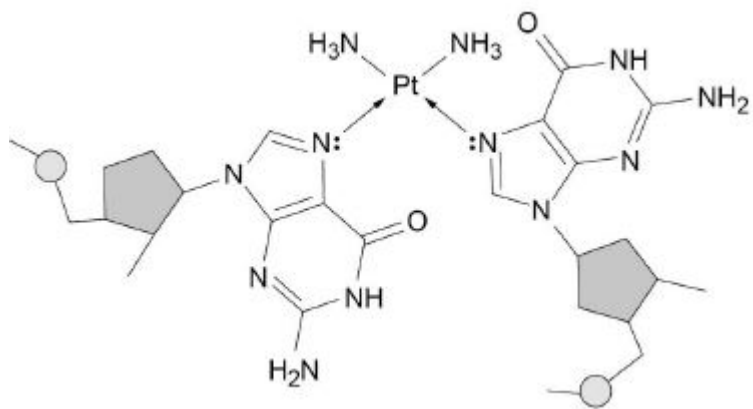
*ALLOW complexes without [ ] and/or ( ) around H<sub>2</sub>O*

*IGNORE ( ) around Cl*

*NOT any additional different species (loses **M2**)*

*(allow uncancelled water on both sides)*

(c)



**M1** Pt in a cis-diammine complex bonded to the correct nitrogen atoms

Pt must have the two ammonia ligands shown

NOT if drawn as trans

IGNORE any charge on Pt

Ignore any wedges and dashes (3D representations)

1

**M2** both lone pairs shown **OR** two arrows indicating co-ordinate bonds

Allow **M2** if bonds to platinum are from the incorrect nitrogen atoms

1

(d) **M1** plot concentration (y-axis) against time (x-axis) **and** take tangents / (calculate the gradients (to calculate rates)

Allow concentration-time graph

NOT time-concentration graph (unless clarified in words or sketch)  
but mark on

1

**M2** Plot rate/gradients against conc

1

**M3** straight line through origin / directly proportional confirms first order  
allow first order if rate halves/doubles when conc halves/doubles

1

Alternatives to **M2** and **M3**:

**M2** Plot a graph of log rate vs log conc

**M3** (Straight) line of gradient = 1

**M2** measure (at least) two half-lives (in this case, tangents not required for M1)

**M3** constant half-life means first order

**M2** compare rates/gradients at different concentrations

**M3** first order if rate halves when conc halves



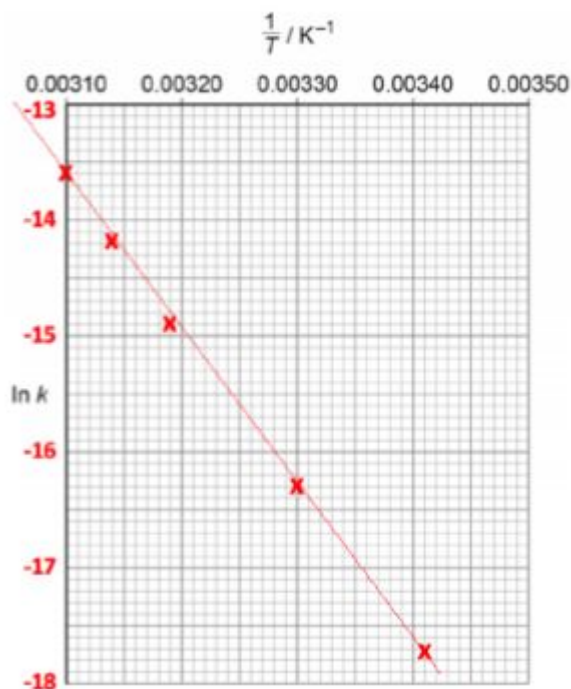
(e)

temperature, $T / \text{K}$	$\frac{1}{T} / \text{K}^{-1}$	rate constant, $k / \text{s}^{-1}$	$\ln k$
318	0.00314	$6.63 \times 10^{-7}$	-14.2

Allow  $3.14 \times 10^{-3}$

2

(f)



Gradient = -13 125

$$\left( -13125 = \frac{-E_a}{R} \right)$$

$$E_a = 13\,125 \times 8.31 = 109\,069 \\ = 109 \text{ (kJ mol}^{-1}\text{)}$$

Vertical axis with sensible scales (plotted points must take up more than half the grid) NOT M1 if y-axis in wrong direction

1

all points plotted correctly (within  $\pm 0.5$  small square)

1

Best fit straight line based on the student's data (ignoring anomalous point if relevant)

1

Gradient calculated within range: 12876 - 13598

1

Mark is for their (gradient  $\times 8.31$ ) **and** conversion into  $\text{kJ mol}^{-1}$

$E_a$  in the range: 107 – 114  $\text{kJ mol}^{-1}$

NOT a negative activation energy

1

[15]

<b>6.</b>	(a) 2-deoxyribose	1
	(b) Base A	
	<i>If Base B stated, allow 1 mark only for response including hydrogen bonding</i>	1
	Top N–H forms hydrogen bonds to lone pair on O of guanine	1
	The lone pair of electrons on N bonds to H–N of guanine	1
	A lone pair of electrons on O bonds to lower H–N of guanine	
	<i>Allow all 4 marks for a correct diagram showing the hydrogen bonding</i>	
	<i>Students could also answer this question using labels on the diagram</i>	1
	(c) Allow either of the nitrogen atoms with a lone pair NOT involved in bonding to cytosine	1
	(d) Use in very small amounts / target the application to the tumour	1
		<b>[7]</b>
<b>7.</b>	(a) $\text{Pt}(\text{NH}_3)_2\text{Cl}_2 + \text{H}_2\text{O} \rightarrow [\text{Pt}(\text{NH}_3)_2\text{Cl}(\text{H}_2\text{O})]^+ + \text{Cl}^-$	
	Correct product	1
	Balanced equation	1
	(b) (i) Hydrogen bond	1
	Oxygen (or nitrogen)	
	<i>Only score this mark if type of bond is correct</i>	1
	(ii) Co-ordinate	1
	Nitrogen (or oxygen)	
	<i>Bond type must be correct to score this mark but allow M2 if bond is covalent</i>	1

- (c) Killing them or causing damage (medical side effects)  
*Allow any correct side effect (e.g. hair loss)*  
*Allow kills healthy (or normal) cells*

1

May attach to DNA in normal cells

1

**[8]**