



A-Level Chemistry

Hydrogen NMR

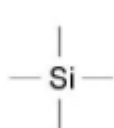
Mark Scheme

Time available: 62 minutes

Marks available: 59 marks

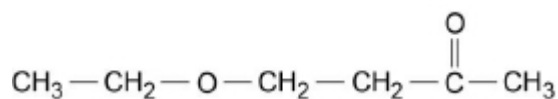
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Mark schemes

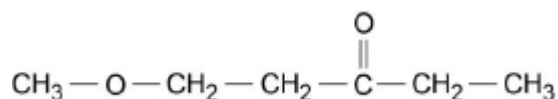
1.	(a) Tick in carbonyl box only	1
	(b) Peak at 2220-2260 cm^{-1} (for $\text{C}\equiv\text{N}$) disappears <i>If both $\text{C}\equiv\text{N}$ disappears and N-H appears without wavenumbers scores 1</i>	M1
	Peak at 3300-3500 cm^{-1} (for N-H) appears	M2
	Fingerprint region different	M3
	(c) Integration ratio 2:2:3 <i>If no link between delta value and oxygen and chlorine, then can award 1 mark for correct explanation of splitting of all 3 peaks</i>	M1
	Peak at 3.95 triplet (integration 2) Cl- CH_2 next to CH_2	M2
	Peak at 3.65 triplet (integration 2) O- CH_2 next to CH_2 <i>If no explanation of splitting, then can award 1 mark for 3 correct links between delta value and oxygen and chlorine M1</i>	M3
	Peak at 3.35 singlet (integration 3) O- CH_3 no adjacent H	M4
	Structure $\text{CH}_3\text{-O-CH}_2\text{CH}_2\text{Cl}$	M5
		[9]
2.	(a)	
		
	1xA01	1
	(b) S	1
	R	1
	Q	1

- (c) (Isomer T)
 signals due to OH (alcohol) at 3230–3350 and C=O at 1680–1750 1
- OH and C=O (functional groups) separated in molecule.
Allow not a carboxylic acid. 1
- (Isomer U)
 (only) signal for OH (alcohol) at 3230–3350 1
- 2 × OH groups present / diol / OH & cyclo(ether) structure.
Allow OH but not C=O. 1
- (Isomer V)
 signals due to OH (acid) at 2500–3000 (and C=O at 1680–1750)
 carboxylic acid group / –COOH present. 1
1
- (d) 2:2:2:3:3
Any order. 1
- (e) (The quartet at $\delta=3.5$ is for a CH₂ group) next to –O–CH₂ **OR** shifted significantly downfield by electronegative O 1
- (is a quartet) because of an adjacent CH₃ group / couple with 3 adjacent protons 1
- (singlet at $\delta=2.2$ is for a CH₃ group) attached to $\begin{array}{c} \text{O} \\ || \\ \text{---C---} \end{array} \text{CH}_3$ **OR** shifted downfield by electronegative C=O 1
- (is a singlet) because there are no adjacent protons / no coupling. 1

(f)



Allow 1 mark for:



2

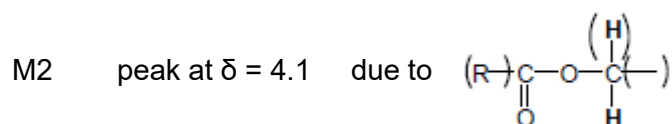
[17]

3.

(a) M1 Ester 1

If Ester 2, can score M3 only.

1



When marking M2 and M3, check any annotation of structures in the stem at the top of the page.

1

M3 ($\delta = 4.1$ peak is) quartet as adjacent / next to / attached to CH_3

1

M4 Other spectrum quartet at $\delta = 2.1-2.6$ (or value in this range)

1

(b) M1 Quaternary (alkyl) ammonium salt / bromide

1

M2 CH_3Br or bromomethane

Penalise contradictory formula and name.

1

M3 Excess (CH_3Br or bromomethane)

Mention of acid eg H_2SO_4 OR alkali eg NaOH loses both M2 and M3.

1

M4 Nucleophilic substitution

Can only score M3 if reagent correct.

Ignore alcohol or ethanol (conditions) or Temp.

1

(c)

	Bromine (penalise Br but mark on)	Acidified KMnO_4 (Penalise missing acid but mark on)
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Wrong reagent = no marks.

If bromine colour stated it must be red, yellow, orange, brown or any combination, penalise wrong starting colour.

1

Benzene	no reaction / colour remains / no (visible) change	no reaction / colour remains / no (visible) change
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Ignore 'clear', 'nothing'.

Allow colour fades slowly.

Allow 'nvc' for no visible change.

1


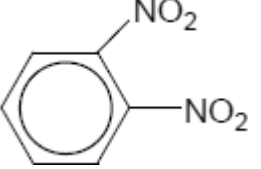
cyclohexene	(Bromine) decolourised	(Acidified KMnO_4) decolourised
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1

[11]

4.

(a)

F	G
	

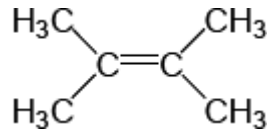
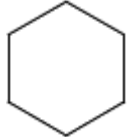
Penalize $-\text{O}_2\text{N}$ once

Penalise missing circle once

Don't penalise attempt at bonding in NO_2

1

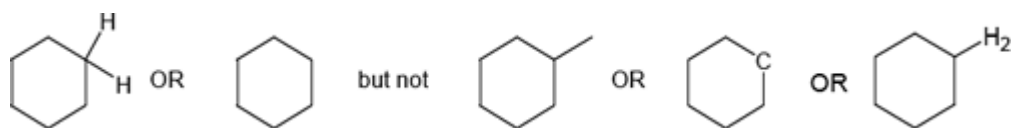
(b)

H	J
	

If both H and J correct but reversed, award one mark

1

A carbon in saturated ring structures should be shown as



(c)

K	L
OR 	
OR 	

1

(d)

M	N
OR 	

Allow C_2H_5 but
 NOT allow C_4H_9 or C_3H_7

1
1

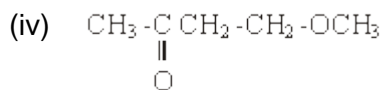
[8]

(ii) $\text{H}_3\text{C}-\text{O}$ or ROCH_3 ;
 (allow 1 if both (i) and (ii) give CH_3- or $\text{H}_3\text{C}-$ only)

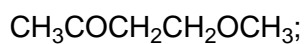
1

(iii) CH_2CH_2 or two adjacent methylene groups;

1

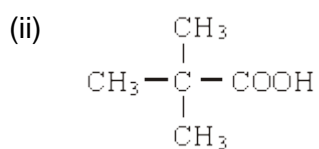


OR



1

(b) (i) OH in acids or (carboxylic) acid present



(c)

reagent	$\text{K}_2\text{Cr}_2\text{O}_7 / \text{H}^+$	$\text{KMnO}_4 / \text{H}^+$
Y	no reaction	no reaction
Z	orange to green or turns green	purple to colourless or turns colourless

5

[9]