

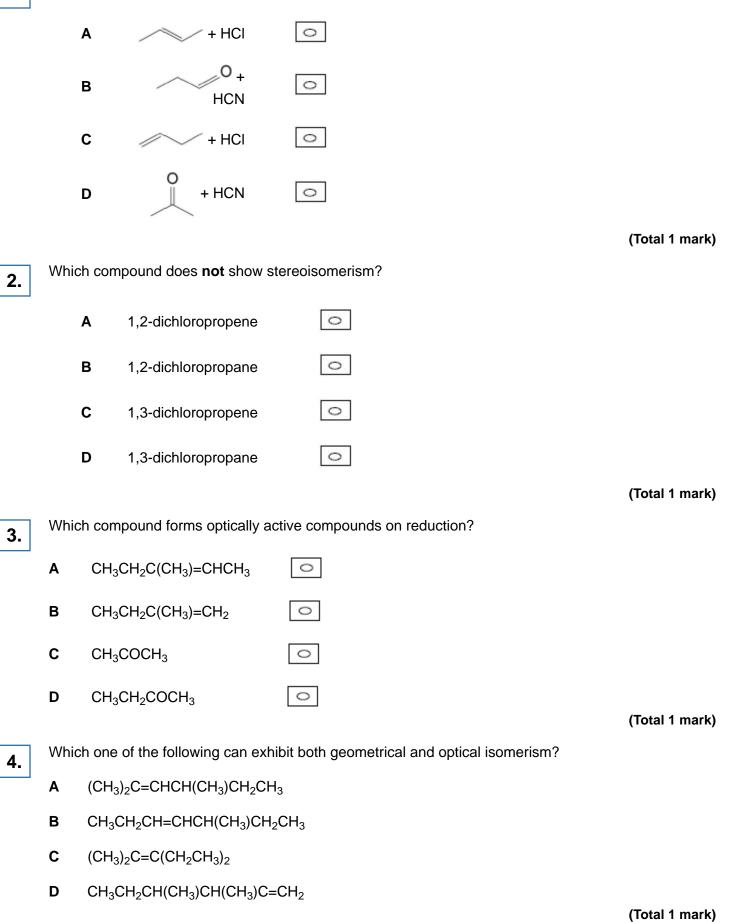
A-Level Chemistry

Optical Isomerism (Multiple Choice)

Question Paper

Time available: 9 minutes Marks available: 9 marks

www.accesstuition.com



Which one of the following reactions will produce an organic compound that has optical isomers?

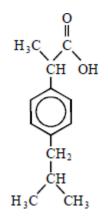
- A dehydration of butan-2-ol by heating with concentrated sulphuric acid
- **B** reduction of pentan-3-one by warming with NaBH₄
- **C** addition of Br₂ to 3-bromopropene

5.

D reduction of 2,3-dimethylpent-2-ene with H₂ in the presence of a nickel catalyst

(Total 1 mark)

6. Ibuprofen is a drug used as an alternative to aspirin for the relief of pain, fever and inflammation. The structure of ibuprofen is shown below.



Which one of the following statements is not correct?

- A It has optical isomers.
- **B** It liberates carbon dioxide with sodium carbonate solution.
- **C** It undergoes esterification with ethanol.
- **D** It undergoes oxidation with acidified potassium dichromate(VI).

(Total 1 mark)

Which one of the following statements about but-2-enal, CH₃CH=CHCHO, is **not** true?

A It has stereoisomers.

7.

- **B** It shows a strong absorption in the infra-red at about 1700 cm⁻¹.
- **C** It will turn an acidified solution of potassium dichromate(VI) green.
- **D** It can be dehydrated by concentrated sulphuric acid.

(Total 1 mark)

- On reduction, a racemate can be formed by
 - A CH₃CH₂CH₂CH₂CHO

8.

- B CH₃CH₂CH₂COCH₃
- C CH₃CH₂COCH₂CH₃
- **D** CH₃CH=CHCH₂CHO

(Total 1 mark)

9. Which one of the following reaction mixtures would give a product capable of exhibiting optical isomerism?

- **A** $CH_3CH=CH_2$ + HBr
- **B** $CH_3CH_2CH_2Br$ + NaOH
- **C** $CH_3CH_2CH_2OH$ + H_2SO_4
- **D** $CH_3CH_2 CHO + HCN$

(Total 1 mark)