

- M1.(a) (i) Appreciates pV should be constant for isothermal change (by working or statement) $W = p\Delta V$ is TO

*Allow only products seen where are approximately 150 for 1 mark
Penalise J as unit here*

M1

Demonstrates $pV = \text{constant}$ using 2 points (on the line) set equal to each other or conclusion made or **shows** that for V doubling that p halves (worth 2 marks)

need to see values for p and V

*Products should equal 150 to 2 sf
Accept statement that products are slightly different so not quite isothermal*

A1

Demonstrates $pV = \text{constant}$ using 3 points (on the line) with conclusion

Need to see values for p and V

*Products should equal 150 to 2 sf
Accept statement that products are slightly different so not quite isothermal*

A1

3

- (ii) Adiabatic therefore no heat transfer **or**
Adiabatic therefore $Q = 0$

B1

Work is done by gas therefore W is negative **or**
Work is done by gas therefore energy is removed from the system

B1

ΔU is negative therefore internal energy of gas decreases **or** energy is removed from the system therefore internal energy of gas decreases or work done by the gas so internal energy decreases

Allow

$$-\Delta U = -W \text{ or } \Delta U = -W$$

B1

3

- (iii) Uses $pV / T = \text{constant}$ or uses $pV = nRT$ or uses $pV = NkT$

e.g. makes T subject or substitutes into an equation with p_A and V_A or p_C and V_C (condone use of $n = 1$) or

their $\frac{(pV)_A}{(pV)_C}$

V_A read off range

$$= 2.5 \text{ to } 2.6 (\times 10^{-4})$$

$$p_A = 600 \times 10^3$$

V_C read off range

$$= 8.5 \text{ to } 8.6 (\times 10^{-4})$$

$$p_C = 140 \times 10^3$$

C1

Correct substitution of coordinates (inside range) into

$$\frac{(pV)_A}{(pV)_C}$$

With consistent use of powers of 10

$(pV)_A$ range is 150 to 156 and $(pV)_C$ range is 119 to 120.4

C1

1.2(5) Allow range from 1.2 to 1.3

Accept decimal fraction : 1

A1

3

- (b) Energy per large square = 10(J) **or** states that work done is equal to area under curve (between A and B)
or energy per small square = 0.4(J)
or square counting seen on correct area

Must be clear that area represents energy either by subject of formula or use of units on 10 or 0.4

Alternative:

W = area of a trapezium

(with working)

or $W = P_{\text{mean}} \times \Delta V$ **or**

$$W = 450 \times 10^3 \times 2.5 \times 10^{-4}$$

or *W = area of a rectangle + area of a triangle (with working)*

B1

Number of large squares = 10.5 to 11.5 seen and (W) =
number of squares \times area of one square (using numbers)
Range = 105 to 115 (J)
Or

Number of small squares = 263 to 287 seen and (W) =
number of squares \times area of one square (using numbers)
Range = 105 to 115 (J)
*States that actual work done would be lower
because of curvature of line*

B1

2

- (c) (Total energy removed per s =) 4560 (J)
or number of cycles per s = 40
or (Mass per second =) $114 \div 68400$ in rearranged form
or their energy $\div (c \Delta T)$ or their energy $\div 68400$

C1

0.067 (kg) seen Allow 0.066 (kg) here
or allow $V / t = 1.67 \times 10^{-3} \div 1100$

or $\left(\frac{V}{t}\right) = \frac{E}{\rho c \Delta \theta}$ and correct **substitution** seen

Condone $E = 114$ (J) or temperature = 291(K)

C1

= 0.061×10^{-3} or 6.06×10^{-5} (m^3)

A1

3

[14]

- M2.(a)** (i) Clear statement that for isothermal $pV = \text{constant}$ or $p_1V_1 = p_2V_2$ ✓
Applies this to any 2 points on the curve AB ✓
e.g. $1.0 \times 10^5 \times 1.2 \times 10^{-3} = 4.8 \times 10^5 \times 0.25 \times 10^{-3}$ $120 = 120$

*Allow $pV = c$ applied to intermediate points **estimated** from
graph e.g. $V = 0.39 \times 10^{-3}$, $p = 3 \times 10^5$*

2

(ii) $W = p \Delta v$
 $= 4.8 \times 10^5 \times (0.39 - 0.25) \times 10^{-3}$
 $= 67 \text{ J } \checkmark$

1

(b)

| | Q / J | W / J | ΔU / J | |
|---------------------------|-------|-------|----------------|--------------|
| process A \rightarrow B | -188 | -188 | 0 | \checkmark |
| process B \rightarrow C | +235 | (+)67 | (+)168 | \checkmark |
| process C \rightarrow A | 0 | +168 | -168 | \checkmark |
| whole cycle | +47 | +47 | 0 | \checkmark |

Any horiz line correct up to max 3
Give CE in B \rightarrow C if ans to ii used for W
If no sign take as +ve

max 3

(c) $\eta_{\text{overall}} = 47 / 235 = 0.20$ or 20% \checkmark

1

- (d) *Isothermal process would require engine to run very slowly / be made of material of high heat conductivity \checkmark*
Adiabatic process has to occur very rapidly / require perfectly insulating container / has no heat transfer \checkmark
Very difficult to meet both requirements in the same device \checkmark
Very difficult to arrange for heating to stop exactly in the right place (C) so that at end of expansion the curve meets the isothermal at A \checkmark

Do not credit bald statement to effect
adiabatic / isothermal process not possible - must give reason

Ignore mention of valves opening / closing, rounded corners, friction, induction / exhaust strokes

wtte

max 2

[9]

M3. (a) (i) Indicated work per cylinder = area of loop ✓ [either stated explicitly or shown on the Figure e.g. by shading or ticking squares or subsequent correct working.]

appropriate method for finding area e.g. counting squares ✓

correct scaling factor used [to give answer of 470 J ± 50 J] ✓

indicated power = $4 \times 0.5 \times (4100/60) \times 470$

= 64 kW ✓

4

(ii) (Fuel flow rate = $0.376/100 = 0.00376 \text{ litre s}^{-1}$)

Input power (= c.v. × fuel flow rate)

= $38.6 \times 10^6 \times 0.00376$ ✓

(= 145 kW)

$\eta_{\text{overall}} = \text{brake power}/\text{input power}$ ✓ seen or implied from correct subsequent working

= $55.0/145 = 0.38$ or 38% ✓

3

(b) Power expended in overcoming friction

in (all) the bearings / between piston & cylinder ✓

and / or in circulating oil / cooling water ✓

and / or driving auxiliaries (e.g. fuel injection pump) ✓

1

(c) Represents the induction and exhaust (strokes) (which take place at nearly atmospheric pressure). ✓

1

[9]

M4. (a) (i) work done (per kg) = area enclosed (by loop) (1)
suitable method of finding area (e.g. counting squares) (1)

correct scaling factor (1)
(to give answer ≈ 500 kJ)

(ii) P (= work done per kg \times fuel flow rate)
 $= 500$ (kJ) $\times 9.9$ (kgs⁻¹) = 5000kW (1)
(4950kW)

(iii) (output power = indicated power – friction power)
 $P_{out} = 4950 - 430 = 45(20)$ kW (1)
(use of $P = 5000$ gives $P_{out} = 45(70)$ kW)
(allow C.E. for values of P in (ii))

5

(b) (i) P_{in} (= fuel flow rate \times calorific value)
 $= 0.30 \times 44 \times 10^6 = 13(2) \times 10^6$ W (1)

$$\text{efficiency} = \frac{4520 \times 10^3}{13.2 \times 10^6} = 34\% \quad (1)$$

(allow C.E. for value of P_{out} in (a) (iii) and P_{in} in (b) (i))

2

[7]

M5. (a) $T_H = 273 + 820 = 1093$ (K), $T_C = 273 + 77 = 350$ (K) (1)

$$\text{efficiency} = \frac{T_H - T_C}{T_H} = \frac{1093 - 350}{1093} = 0.68 \text{ or } 68\% \quad (1)$$

2

(b) rotational speed of output shaft = $\frac{1800}{2 \times 60} = 15$ rev s⁻¹ (1)

(work output each cycle = 380 J, 2 rev \equiv 1 cycle in a 4 stroke engine)

$$\text{indicated power} = 15 \times 190 = 5.7 \text{ kW (1)}$$

2

(c) $\text{power lost (= indicated power - actual power)} = 5.7 - 4.7 = 1.0 \text{ kW (1)}$
(allow C.E. for incorrect value from (b))

1

(d) $\text{energy supplied per sec (= fuel flow rate} \times \text{calorific value)}$

$$= \frac{2.1 \times 10^{-2}}{60} \times 45 \times 10^6 = 16 \text{ kW (15.8 kW) (1)}$$

1

(e) $\text{efficiency} = \frac{\text{net power output}}{\text{power input}} = \frac{4.7}{16} = 0.29 \text{ or } 29 \%$

$$\frac{4.7}{15.8} = 0.30 \text{ or } 30\%$$

(allow C.E. for value from (d))

1

[7]