

- M1.(a)** (i) electromagnetic / electrostatic / Coulomb (repulsion between the alpha particles and the nuclei) ✓

The interaction must be named not just described.

1

- (ii) the scattering distribution remains the same (because the alpha particles interact with a nucleus) whose charge / proton number / atomic number remains the same or the (repulsive) force remains the same

The mark requires a described distribution and the reason for it.

Or

the scattering distribution changes / becomes less distinct because there is a mixture of nuclear masses (which gives a mixture of nuclear recoils)

✓

(owtte)

A reference must be made to mass and not density or size.

1

- (b) (i) use of graph to find r_0
 e.g. $r_0 = 6.0 \times 10^{-15} / 75^{1/3}$ ✓
 (or $8.0 \times 10^{-15} / 175^{1/3}$)
 ($r_0 = 1.43 \times 10^{-15}$ m)

Substitution and calculation t must be shown.

Condone a gradient calculation on R against $A^{1/3}$ graph (not graph in question) as $R \propto A^{1/3}$

1

- (ii) **Escalate if clip shows ²⁷13Al in the question giving $R \approx 4 \times 10^{-15}$ m.**

(using $R = r_0 A^{1/3}$)

$$R = 1.43 \times 10^{-15} \times 51^{1/3} \quad \checkmark$$

$$R = 5.3 \times 10^{-15} \text{ (m)} \quad \checkmark$$

$$(R = 5.2 \times 10^{-15} \text{ m from}$$

$$r_0 = 1.4 \times 10^{-15} \text{ m})$$

First mark for working.

Second mark for evaluation which must be 2 or more sig figs allow CE from (i) $R = 3.71 \times (i)$.

Possible escalation.

2

- (c) Escalate if clip shows ²⁷13 in the question and / or the use of 27 in the working.

density = mass / volume

$$m = 51 \times 1.67 \times 10^{-27}$$

$$= 8.5 \times 10^{-26} \text{ kg}$$

Give the first mark for substitution of data into the top line or bottom line of the calculation of density.

$$v = 4/3\pi (5.3 \times 10^{-15})^3$$

$$(6.2(4) \times 10^{-43} \text{ m}^3)$$

In the second alternative the mark for the substitution is only given if the working equation is given as well.

Or

$$\text{density} = A \times u / 4/3\pi (r_0 A^{1/3})^3$$

$$= u / 4/3\pi (r_0)^3$$

51 × 1.67 × 10⁻²⁷ would gain a mark on its own but 1.66 × 10⁻²⁷ would need u / 4/3 π(r₀)³ as well to gain the mark.

$$\text{top line} = 1.66 \times 10^{-27}$$

$$\text{bottom line} = 4/3\pi (1.43 \times 10^{-15})^3$$

✓ for one substitution

$$\text{density} = 1.4 \times 10^{17} \quad \checkmark$$

$$(1.37 \times 10^{17})$$

$$\text{kg m}^{-3} \quad \checkmark$$

Expect a large spread of possible answers. For example if $R = 5 \times 10^{-15}$ $V = 5.24 \times 10^{-43}$ and density = 1.63×10^{17} .

Possible escalation.

3

[8]

- M2.(a)** (i) (Mass change in u) 1.71×10^{-3} (u)
or (mass Be-7) – (mass He-3) – (mass He-4) seen with numbers

C1

$$2.84 \times 10^{-30} \text{ (kg)}$$

or Converts their mass to kg

Alternative 2nd mark:

Allow conversion of 1.71×10^{-3} (u) to MeV by multiplying by 931 (=1.59 (MeV)) seen

C1

Substitution in $E = mc^2$ *condone their mass difference in this sub but must have correct value for c^2 (3×10^8)² or 9×10^{16}*

Alternative 3rd mark:

*Allow their MeV converted to joules ($\times 1.6 \times 10^{-13}$) **seen***

C1

2.55×10^{-13} (J) to 2.6×10^{-13} (J)

Alternative 4th mark:

Allow 2.5×10^{-13} (J) for this method

A1

4

(ii) Use of $E = hc / \lambda$ **ecf**

C1

Correct substitution in rearranged equation with λ *subject ecf*

C1

7.65×10^{-13} (m) to 7.8×10^{-13} (m) **ecf**

A1

3

(b) (i) Use of E_p formula:

C1

Correct charges for the nuclei **and** correct powers of 10

C1

$2.6(3) \times 10^{-13}$ J

A1

3

(ii) Uses $KE = 3 / 2 kT$: **or halves KE_T , $KE = 1.3 \times 10^{-13}$ (J) **seen ecf****

C1

Correct substitution of data **and** makes T subject **ecf**
Or uses KE_T value **and** divides T by 2

C1

6.35×10^9 (K) or 6.4×10^9 (K) or 6.28×10^9 (K) or 6.3×10^9 (K) **ecf**

A1

3

(c) (i) Deuteron / deuterium / hydrogen-2

B1

Triton / tritium / hydrogen-3

B1

2

(ii) Electrical heating / electrical discharge / inducing a current in plasma / use of e-m radiation / using radio waves (causing charged particles to resonate)

B1

1

[16]

M3.D

[1]

M4.(a) the amount of energy required to separate a nucleus ✓
into its separate neutrons and protons / nucleons ✓
(or energy released on formation of a nucleus ✓
from its separate neutrons and protons / constituents ✓)

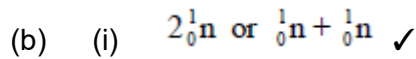
1st mark is for correct energy flow direction

2nd mark is for binding or separating nucleons (nucleus is in the question but a reference to an atom will lose the mark)

ignore discussion of SNF etc

both marks are independent

2



must see subscript and superscripts

1

- (ii) binding energy of U
= 235×7.59 ✓ (= 1784 (MeV))
binding energy of Tc and In
= $112 \times 8.36 + 122 \times 8.51$ ✓
(= 1975 (MeV))
energy released (= $1975 - 1784$) = 191 (MeV) ✓ (allow 190 MeV)

1st mark is for 235×7.59 seen anywhere

2nd mark for $112 \times 8.36 + 122 \times 8.51$ or 1975 is only given if there are no other terms or conversions added to the equation (ignore which way round the subtraction is positioned)

correct final answer can score 3 marks

3

- (iii) energy released
= $191 \times 1.60 \times 10^{-13}$ ✓
(= 3.06×10^{-11} J)
loss of mass (= E / c^2)
= $2.91 \times 10^{-11} / (3.00 \times 10^8)^2$
= 3.4×10^{-28} (kg) ✓

or

- = $191 / 931.5$ u ✓ (= 0.205 u)
= $0.205 \times 1.66 \times 10^{-27}$ (kg)
= 3.4×10^{-28} (kg) ✓

allow CE from (ii)

working must be shown for a CE otherwise full marks can be given for correct answer only

note for CE

answer = (ii) $\times 1.78 \times 10^{-30}$

(2.01×10^{-27} is a common answer)

2

- (c) (i) line or band from origin, starting at 45° up to Z approximately = 20
reading Z = 80, N = 110 → 130 ✓

*initial gradient should be about 1 (ie Z = 20 ; N = 15 → 25)
and overall must show some concave curvature. (Ignore slight waviness in the line)*

if band is shown take middle as the line

if line stops at N > 70 extrapolate line to N = 80 for marking

1

- (ii) fission fragments are (likely) to be above / to the left of the line of stability ✓
fission fragments are (likely) to have a larger N / Z ratio than stable nuclei
or
fission fragments are neutron rich owtte ✓
and become neutron or β^- emitters ✓

*ignore any reference to α emission
a candidate must make a choice for the first two marks
stating that there are more neutrons than protons is not
enough for a mark*

1st mark reference to graph

2nd mark – high N / Z ratio or neutron rich

3rd mark beta minus

note not just beta

3

[12]