Please write clearly in	ı block capitals.		
Centre number		Candidate number	
Surname			
Forename(s)			
Candidate signature			

GCSE SCIENCE A 2

Higher Tier Unit 6

Wednesday 25 May 2016

Afternoon

Time allowed: 1 hour 30 minutes

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Materials

For this paper you must have:

- a ruler
- a calculator
- the Chemistry Data Sheet and Physics Equations Sheet booklet (enclosed).

Instructions

- Use black ink or black ball-point pen.
- Fill in the boxes at the top of this page.
- Answer all questions.
- You must answer the questions in the spaces provided. Do not write outside the box around each page or on blank pages.
- Do all rough work in this book. Cross through any work you do not want to be marked.

Information

- The marks for questions are shown in brackets.
- The maximum mark for this paper is 90.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.
- Question 3(a) should be answered in continuous prose. In this question you will be marked on your ability to:
 - use good English
 - organise information clearly
 - use specialist vocabulary where appropriate.

Advice

• In all calculations, show clearly how you work out your answer.













Turn over ►







Turn over ►

- **3** Polar bears live in the Arctic. Seals are their main food source.
- **3 (a)** Figure 4 shows some adaptations of polar bears.

Figure 4





In this question you will be assessed on using good English, organising information clearly and using specialist terms where appropriate.

Explain how the polar bear is adapted to survive in its environment.

Use information in Figure 4 and your knowledge.

[6 marks]



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Turn over ►



4 (a) A student investigated the energy released by burning the vegetable oil.

Figure 6 shows the apparatus the student used.





The student:

- recorded the starting temperature of the water
- burned 1 g of the vegetable oil
- recorded the temperature of the water after burning the vegetable oil.



	Table 1 shows the student's results for the investigation.			
	Starting temperature Temperature of water after of water in °C burning 1 g of vegetable oil in °C			
	19	34		
4 (a) (i)	Calculate the energy released when Use the equation: Energy released	1 g of the vegetable oil was burned. in joules = 100 \times 4.2 \times temperature of	change [2 marks]	
	Energy released whe	n 1 g of vegetable oil was burned =	J	
4 (a) (ii)	Use your answer to part (a)(i) to calc vegetable oil.	ulate the energy released by 100 g of		
	Convert your answer to kilojoules.		[1 mark]	
	Energy released when 1	00 g of vegetable oil was burned =	kJ	
4 (a) (iii)	The student did not get an accurate vegetable oil. Suggest two reasons	value for the energy released by 100 g why.	of the [2 marks]	
	Question 4 contin	ues on the next page		

Turn over ►

The student compared the cooking of potatoes in vegetable oil and in water.

The potatoes cooked in vegetable oil were a different colour and texture to the potatoes cooked in water. Give two other differences between the potatoes cooked in vegetable oil and those cooked in water. [2 marks] 4 (c) The student added bromine water to the vegetable oil. Which structure turns bromine water from orange to colourless? [1 mark] Tick (\checkmark) one box. н Н O-H-O-H С С C = CC = OH /n Н Suggest one reason why information about energy is shown on food labels. 4 (d) [1 mark]



4 (b)

5	Polymers are used to make plastic bags.
5 (a)	Most plastic bags are made from a polymer produced from ethene.
	What is the name of the polymer produced from ethene?
	[1 mark]
5 (b)	Ethene is produced by cracking hydrocarbons obtained from crude oil.
5 (b) (i)	Complete the symbol equation for the reaction to produce ethene and octane
	[1 mark]
	$C_{10}H_{22} \longrightarrow C_{2}H_{4} + $
5 (D) (II)	Give one use for octane. [1 mark]
5 (c)	Plastic bags in supermarkets used to be free. Supermarkets now make a charge. It is estimated that the number of plastic bags used has dropped by 75%.
5 (c) (i)	When they were free, 8 billion plastic bags a year were used.
	Estimate the number of plastic bags that are now used each year.
	[1 mark]
	Number of plastic bags used each year =
5 (c) (ii)	Suggest two reasons why the government made a law to charge for plastic bags.
	[2 marks]

Turn over ►











6 (c) (i)	The time recorded on the	ne stopwatch is 0.29 se	econds.	
	What is the resolution of the stopwatch?			
	Draw a ring around the	correct answer.		ניוומיאן
	0.1 s	0.01 s	0.001 s	
6 (c) (ii)	Suggest two ways to ob	otain a more accurate i	result	
0 (0) (1)				[2 marks]



Microwaves are a type of electromagnetic wave. Microwaves are used by mobile phones. Give two properties of all electromagnetic waves. 7 (a) (i) [2 marks] 7 (a) (ii) Complete the sentences about electromagnetic waves. [2 marks] Radiowaves have the longest ______ and lowest _____ of all electromagnetic waves. 7 (b) Some scientists are investigating if brain cancer is linked to the use of mobile phones. In one study the scientists analysed information from phone companies from more than 358 000 people who used mobile phones. The study looked for a link between the amount of time people use their mobile phones and the number of people who developed brain cancer. The scientists could not come to a definite conclusion. 7 (b) (i) Give **one** advantage of using phone company records to look for a link, instead of asking people about their mobile phone use. [1 mark] 7 (b) (ii) Suggest one reason why the scientists could not come to a definite conclusion. [1 mark]



7

	Biology Questions		
8	Living organisms can be used as indicators of pollution.		
8 (a) (i)	Which group of organisms can be used to indicate the concentration of dissolved oxygen in water? [1 mark]		
8 (a) (ii)	Lichen species are indicators of a pollutant gas. Name the pollutant gas. [1 mark]		
8 (b)	Scientists investigated the distribution of lichens at different distances to the north and south of a power station.		
	The scientists used a sampling square called a quadrat to estimate the percentage cover of lichens on trees.		
	Each quadrat was divided into 16 smaller squares.		
	Only the small squares in which the lichens covered more than half the square were counted. This number of squares was then used to calculate the percentage cover of lichens in the whole quadrat.		
	Figure 9 shows the lichens growing in one quadrat.		
	Figure 9		
	Lichens		
	Calculate the percentage cover of lichens in the quadrat shown in Figure 9. [1 mark]		
	Percentage cover of lichens = %		







9 The materials in organisms are constantly recycled.

Many trees lose their leaves every year, and most animals produce waste at least once a day. All plants and animals eventually die.

Describe the processes that return the chemicals in waste and in dead organisms to the environment.

[4 marks]

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10	Variation between individuals of the same species can be caused by genes inhoriton their parents.	erited
10 (a)	State the other cause of variation between individuals of the same species.	
	Give one example of this cause, and describe how it could affect the individual. [2	2 marks]
10 (b)	Scientists can alter the genetic make-up of organisms so that they develop des characteristics.	ired
	Recently the treatment of cancers has improved. This is partly due to research genetically engineered mice that develop cancer.	using
	Evaluate the use of mice that develop cancer in medical research.	3 marks]
	Turn over for the part question	
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11 (c)	How is olive oil hardened to form a spread?	[2 marks]
	Turn over for the next question	
		Turn over ▶





- 12 This question is about gases in the atmosphere.
- **12 (a)** The atmosphere is a mixture of gases. The gases can be separated to provide raw materials for industrial processes.

Table 2 shows the melting and boiling points of some gases in the atmosphere.

Gas	Melting point in °C	Boiling point in °C
Carbon dioxide	-78	-78
Neon	-249	-246
Nitrogen	-210	-196
Oxygen	-219	-183

Table 2

The gases in the atmosphere are separated by the following method:

- water vapour and carbon dioxide are removed
- the remaining gases are cooled to –200 °C
- gases that do not condense at -200 °C are removed
- the liquid mixture is separated.
- **12 (a) (i)** Give the full name of the process used to separate the liquid mixture.

12 (a) (ii)	Which gas does not con Draw a ring around the	ndense at –200 ° correct answer.	°C?		[1 mark]
	carbon dioxide	neon	nitrogen	oxygen	
12 (a) (iii)	Suggest why carbon dic cooled to –200 °C.	oxide and water a	are removed before t	the gases are	[1 mark]

Turn over ►



12 (b) Scientists think Earth's early atmosphere was like the atmosphere of Mars today.

 Table 3 shows the percentages of some of the gases in the atmosphere of Mars.

Га	ıb	le	3

Gas	Percentage (%) of total
Carbon dioxide	95.3
Nitrogen	2.7
Oxygen	0.13

Describe how the percentages of oxygen and carbon dioxide in the Earth's atmosphere today have changed from those in the early atmosphere.

Give reasons for these changes.

[3 marks]



12 (c) Some scientists believe life started after gases in Earth's early atmosphere reacted with each other to produce amino acids. Two scientists, Miller and Urey, modelled the early atmosphere and produced amino acids in an experiment.

Figure 11 shows the basic structure of an amino acid.





Turn over ►





13 (b)	A disadvantage of wind turbines is that the wind is an unreliable energy source. Geothermal power stations and hydroelectric power stations use more reliable ener sources.			
	hydroelectric power stations.			
	Suggest reasons why. [3 marks]			
Turn over for the next question				



Turn over ►

Some scientists study cosmic microwave background radiation (CMBR) and the speed

	of galaxies at different distances from	the Earth.		
14 (a) (i)	What is CMBR?			
			[2 marks]	
14 (a) (ii)	How did the discovery of CMBR supp	ort the 'Big Bang' theory?		
			[1 mark]	
14 (b) (i)	A student collected data about the spetter for the Earth. Some of the data that the	eed of galaxies at different	distances from	
Table 4				
	Distance of galaxy	Speed of galaxy		
	from Earth in million light years	in km/s		
	1.63	200		
	2.61	300		
	3.26	920		
	6.52	1090		
		1	I	

















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