

KS3 Science

Changes of State

Mark Scheme

Time available: 35 minutes Marks available: 51 marks

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Mark schemes

1.	(a)		
		liquid to gas	
		• fat melted gas to liquid	
		liquid to solid	
		• water evaporated solid to liquid	
		solid to gas	
		if more than one line is drawn from either 'fat melted' or 'water evaporated', award no mark for that statement	
			2 (L3)
	(b)	(i) • All fuels are sources of energy.	
		if more than one box is ticked, award no mark	1 (L3)
		(ii) • oxygen 🗸	
		if more than one box is ticked, award no mark	1 (L4)
	(c)	● It is rigid. ✓	
		 It has a very high melting point. ✓ 	
		if more than two boxes are ticked, deduct one	
		mark for each incorrect tick minimum mark zero	
			2 (L4)

[6]

(a)	•	
]	solid to liquid evaporating	
	liquid to gas melting	
	gas to liquid condensing	
	liquid to solid freezing	
	award three marks for all four correct lines	
	award two marks for any three correct lines	
	award one mark for any two correct lines	
	if more than one line is drawn from any change of state, do not credit that change of state	
	-	3 (L3)
(b)	• 0°C	
(U)	accept 'zero'	
	do not accept 'nothing'	
	•	1 (L4)
(c)	it decreased	
(-)	accept 'it got colder'	
	'it dropped to below 0°C' is insufficient	
	any references to time are insufficient	
		1 (L3)
(d)	 Sand increases friction between car tyres and the road. 	
	 Salt makes ice melt. ✓ 	
	if more than two boxes are ticked, deduct a	
	mark for each incorrect box	
	minimum of zero	2 (L4)
		(/

2.

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3.	(a)	
		It does not rust.
		base of a saucepan
		It is a good conductor
		coin of electricity
		It is a good conductor
		of heat.
		wires in a cable
		It is not magnetic.
		if more than one line is drawn from an object,
		award no mark for that object
	(b)	 Brass does not bend as easily as copper.
		if more than two boxes are ticked,
		deduct one mark for each incorrect tick
		 Brass is harder than copper
		minimum mark zero
	(c)	 liquid zinc and solid copper
		if more than one box is ticked, award no mark
	(\mathbf{a})	• weter: increased er went up
4.	(a)	water: increased or went up accept 'expanded'
		accept expanded accept a reading greater than 7 (cm^3)
		and less than or equal to 8 (cm^3)
		 wax: decreased or went down
		accept 'contracted'
		accept a reading lower than 7 (cm^3) and greater
		than or equal to 5.5 (cm ³) both answers are required for the mark
	(b)	• <i>water</i> . liquid
	. ,	•

3 (L3)

1 (L3)

1 (L3)

1 (L4)

1 (L3)

1 (L4)

1 (L4)

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wax: solid

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- (c) (i) any **one** from
 - the liquid had turned blue accept 'the colour changed' accept 'the water was blue or coloured or grey' 'it formed a solution' is insufficient
 - crystals or copper sulphate or solid could not be seen accept 'the crystals disappeared'
 - there was no solid left
 1 (L3)

(ii) any **one** from

- stir it accept 'shake it' **or** 'mix it'
- heat it
 - use warm water accept 'use hotter water' accept 'add more water' do **not** accept 'put less in'

(d) (i) it evaporated accept 'it went into the air'

(ii) copper sulphate accept 'crystals' accept 'sulphate'

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1 (L3)

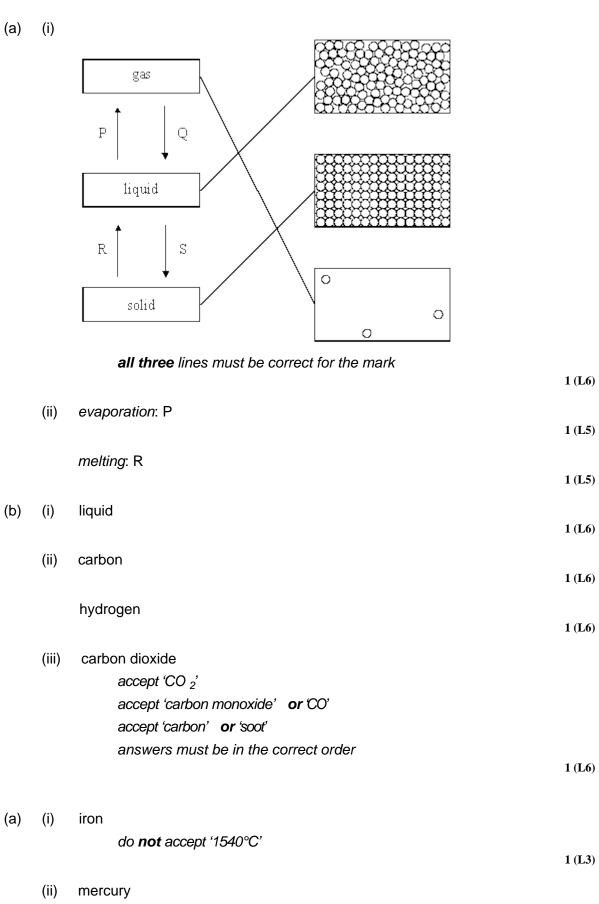
1 (L4)

1 (L3)

] (

5.

6.



do not accept '-37°C'

1 (L3)

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	(b)	solid <i>to a</i> l			
			answers must be in the correct order		
			both answers are required for the mark	1 (L3)	
	(c)	5		1 (L3)	
	(d)	(i) sodi	um	1 (L3)	
		<i></i>			
		(ii) gold		1 (L3)	[6]
7.	(a)	any one fr	om		
		• a dip			
		a hollow	N		
			accept 'a hole'		
		a dent			
			accept a drawing of a hollow		
			do not accept 'it gets thinner'		
				1	
	(b)	it melted			
			accept 'because the air or water was warm'		
				1	
	(c)	a bar at 3	minutes showing a height of between 10 and 8 cm		
				1	[2]
					[3]
8.	(a)	(i) 100			
			accept answers from 98 to 100		
				1 (L5)	
		(ii) to co	ondense the water vapour		
			accept 'to change the gas into liquid'		
			or 'condensation' or 'condenser'		
			accept 'to cool the vapour into water'		
			do not accept 'to cool the vapour or water'	1 (L5)	
		/		- (
		(iii) disti	llation or distilling	1 (L5)	
				- ()	

(ii)

1 (L6)
1 (L6)
1 (L6)
1 (L6)
1 (L6)

0)

[9]