

# **KS3 Science**

## Combustion

**Mark Scheme** 

### Time available: 30 minutes Marks available: 44 marks

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### Mark schemes

1.	(a)	a ruler     accept 'a metre rule'	
		accept 'a tape measure'	
		<i>: 'cm' is insufficient 'a measuring stick' is insufficient</i>	
			1 (L3)
	(b)	(i) • 30	
	(0)	do <b>not</b> accept '30 seconds'	
			1 (L3)
		(ii) • A to B: any number from 5 to 15	
		accept a range such as '5 to 10'	
			1 (L4)
		• D to E: any number from 45 to 80	
		accept a range such as '50 to 60'	1 (L4)
			1 (114)
	(c)	any <b>one</b> from	
		you can measure smaller intervals of time	
		accept 'each section burns for a shorter time'	
		accept 'it is more precise <b>or</b> accurate'	
		'it is easier to read' is insufficient	
		the lines are closer	
		accept 'the lines are close'	
		accept 'the lines are further apart on candle 1' accept 'the lines are 1 cm apart on candle 1	
		and 0.5 cm apart on candle 3'	
		accept 'there are more lines' <b>or</b> smaller spaces	
		or smaller segments'	
		accept 'more sections <b>or</b> rings'	
		'the lines are smaller' is insufficient	
			1 (L4)
	(a)	ethanol <b>or</b> alcohol	
2.	(a)	if more than one box is ticked, award no mark	
		II THOLE THAT ONE DOX IS TICKED, AWAID THO THATK	1 (L3)
			= (==)

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#### (b) any **one** from

3.

	<ul> <li>burning hydrogen does not produce carbon monoxide accept 'petrol or ethanol or alcohol produces carbon monoxide'</li> </ul>						
	burning hydrogen does not produce sulphur dioxide     accept 'petrol produces sulphur dioxide'						
	<ul> <li>burning hydrogen only produces water</li> </ul>						
	<ul> <li>burning petrol causes acid rain accept 'hydrogen or ethanol or alcohol does not cause acid rain'</li> </ul>	1 (L4)					
(c)	hydrogen accept 'H <sub>2</sub> " accept 'gas'	1 (L4)					
(d)	oxygen ✔ if more than one box is ticked, award no mark	1 (L4)					
(e)	any <b>one</b> from						
	• it can be grown accept 'it does not take long to grow'						
	it can be replanted     accept 'it can be replaced'						
	it is renewable						
	it can be reproduced <i>accept 'it produces seeds'</i>	1 (L4)					
(a)	(i) water accept 'H <sub>2</sub> O'	1 (L5)					
	(ii) carbon dioxide accept 'CO <sub>2</sub> '	1 (L5)					

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	(b)	(i)	do not use antifreeze <b>or</b> methanol near a naked flame and do not swallow accept 'it catches fire easily and it is poisonous' accept 'wash hands after use' for do not swallow accept 'it is flammable <b>or</b> inflammable and it is poisonous' <b>both</b> answers are required for the mark	1 (L5)
		(ii)	any <b>one</b> from	
			water froze	
			the mixture froze	
			<ul> <li>the contents froze         <ul> <li>accept '10% antifreeze is not enough to stop the water freezing'</li></ul></li></ul>	1(16)
				1 (L6)
			and expanded	1 (L6)
4.	(a)	tar		1 (L5)
	(b)	(i)	any <b>one</b> from	
			<ul> <li>to condense the vapour accept 'energy is transferred from the water vapour to the ice'</li> </ul>	1 (L5)
		(ii)		
			a random arrangementa random arrangementof particles most of whichof particles most ofdo not touchwhich touch each other	2 (L6)
	(c)			
			accept 'CO 2'	1 (1 =
				1 (L5)

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- 5.
- (a) chemical

any two from

sound

thermal

accept 'potential' or 'stored'

1 (L6)

2 (L6)

2 (L6)

1 (L6)

kinetic

accept 'heat'

- accept 'movement'
- light

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- (b) any **two** from
  - they gained energy
     *accept 'they move more quickly'*
  - they hit the lid with greater force accept 'they hit the lid harder'
  - they hit the lid more often
    - accept 'the pressure inside the tin increased' accept 'the molecules are closer together' accept 'more molecules are present'
- (c) (i) oxygen

accept 'O 2'

- (ii) any **one** from
  - carbon dioxide accept 'CO 2'
  - water vapour accept 'H 2O' accept 'carbon monoxide'

1 (L6)

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- (d) any one from
  - it was quieter
  - the lid didn't move as high accept 'the lid was not pushed off'
  - less energy released accept 'it does not work'
- (a) (i) any **two** from

6.

- CO<sub>2</sub> do **not** accept 'carbon dioxide'
- H<sub>2</sub>O
   do not accept 'water'
- CO
   do not accept 'carbon monoxide'
- C do **not** accept 'carbon'
- (ii) any two from
  - water droplets form on the inside of the bell-jar accept 'condensation'
  - thermal energy is released accept 'heat or energy is given off'
  - light is released
     accept 'there is a flame'
  - soot is produced

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smoke is produced accept 'the wick is used up' do **not** accept 'carbon dioxide is formed' **or** 'carbon monoxide is given off'

2 (L7)

1 (L5)

2 (L7)

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	(b)	(i) a	answers may be in either order				
		•	oxygen increases				
				1 (L7)			
		•	carbon dioxide decreases				
				1 (L7)			
		(ii) p	photosynthesis stops	1 (L7)			
				I (L7)			
		r	respiration continues to take place				
			do <b>not</b> accept 'respiration takes place'	1 (L7)			
				I (L7)			
	(c)	(i) c	chloroplast	1 (1 8)			
				1 (L7)			
		(ii) r	nucleus				
				1 (L7)			
	<i>.</i>						
7.	(a)	any <b>or</b>	ne from				
		• use	ed up in burning				
			accept 'oxygen burned'				
			do <b>not</b> accept 'used up'				
		• it re	eacted with fuel <b>or</b> petrol				
		• form	med carbon dioxide <b>or</b> carbon monoxide				
			accept 'it has been turned into carbon dioxide'				
				1 (L6)			
	(b)	(i) v	water				
			accept 'steam' <b>or</b> 'hydrogen oxide'				
			do <b>not</b> accept 'hydroxide'				
				1 (L6)			
		(ii) it	t is poisonous <b>or</b> toxic				
			accept a reference to its being carried				
			by the blood in preference to oxygen				
			accept 'it can form an explosive mixture with air'				
			do <b>not</b> accept 'it damages the lungs' <b>or</b> 'it is harmful'	1(16)			
				1 (L6)			

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(b)

- (a) any one from
  - more air or oxygen

accept 'gas reaching the flame already has air **or** oxygen mixed in it'

better mixing gives more combustion or more efficient burning • accept 'better or faster combustion' accept the converse i.e. arguments applied to a Bunsen with a closed air-hole 1 (L6) oxygen do not accept 'air' 1 (L6) carbon dioxide + water both products are required for the mark products may be in either order accept 'carbon monoxide + water' disregard any reference to heat or energy accept correct formulae for words the equation need not be balanced

1 (L6)